







Comparative effects of flavonoids on oxidant scavenging and ischemia-reperfusion injury in cardiomyocytes

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Received 6 December 2006; received in revised form 14 March 2007; accepted 19 March 2007 Available online 30 March 2007

Abstract

Since flavonoids scavenge reactive oxygen species, they may potentially protect against ischemia/reperfusion injury. This study compared the scavenging capacity of specific flavonoids towards different reactive oxygen species. Whether the differential oxidant scavenging capacity correlated with their protective efficacy in ischemia/reperfusion injury of cardiomyocytes was determined. The free radical scavenging capacity of five flavonoids (wogonin, baicalin, baicalein, catechin and procyanidin B_2) was analyzed using electron spin resonance spectrometry for 3 radicals: 1,1-diphenyl-2picrylhydrazyl (DPPH), superoxide and hydroxyl radical. A well-established chick cardiomyocyte model of ischemia (1 h)/reperfusion (3 h) was used to evaluate flavonoid-induced protection against ischemia/reperfusion injury in chronic treatment (pretreated 72 h and treated through ischemia/reperfusion) and acute treatment protocols (during ischemia/reperfusion or only at reperfusion). The cell viability was assessed by propidium iodide. The DPPH scavenging was most significant with catechin, followed by procyanidin B_2 , baicalein, baicalin, and wogonin. The superoxide scavenging was, similarly, most significant with catechin, followed by baicalein, procyanidin B_2 , and baicalin. For hydroxyl radical, only baicalein showed a significant scavenging capacity (>50% reduction in ESR signal). For the cardiomyocyte studies, all flavonoids but wogonin showed protection against ischemia/reperfusion injury in the chronic treatment protocol. When flavonoids were administered only during ischemia/reperfusion, baicalein, procyanidin B_2 , and catechin significantly reduced cell death. If flavonoids were administered just at reperfusion, only baicalein and procyanidin B_2 had protective effects, and the efficacy was less. Flavonoids possess specific but differential radical scavenging capacity, which, in conjunction with the timing of treatment, affects their protective efficacy in cardiomyocytes exposed to ischemia/reperfusion.

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Keywords: Flavonoids; Reperfusion injury; Scavenging capacity; Superoxide; Hydroxyl radical

1. Introduction

Flavonoids are naturally occurring polyphenolic compounds found ubiquitously in vegetables and fruits. These compounds demonstrate potent antioxidant effects in a variety of *in vitro* and *in vivo* models using endogenous or exogenous oxidant stress (Gao et al., 1999; Haramaki et al., 1994; Sakano et al., 2005; Wang et al., 2006b). Specifically, flavonoids have been investigated for their ability to prevent oxidant injury caused by

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reperfusion of ischemic tissue, a scenario similar to clinical conditions such as stroke and myocardial infarction. Flavonoids have been typically administered as a pretreatment for several days and a delayed antioxidant protective effect in an ischemia/reperfusion model has been evaluated (Pataki et al., 2002; Sato et al., 2001). Although chronic treatment with flavonoids has shown induction of enhanced antioxidant protection, it may not always be feasible to administer flavonoids as a pretreatment. Therefore, protective role of flavonoids in an acute treatment model of ischemia/reperfusion, i.e. administration of flavonoids concurrently with ischemia/reperfusion or just at the point of reperfusion, is of greater interest.

The type of reactive oxygen species generated during ischemia and reperfusion have been demonstrated to be phasedependent in our cardiomyocyte model of ischemia/reperfusion (Becker et al., 1999; Lebuffe et al., 2003; Vanden Hoek et al., 1997). Superoxide anion is mainly produced during ischemia or hypoxemia as a result of increased electron leak from the impaired mitochondrial electron transport chain. Upon reperfusion, mitochondrial respiratory chain is overwhelmed with the increased oxygen tension from hypoxic to normoxic conditions and results in production of massive quantity of superoxide from complex III of the mitochondrial electron transport chain (Zweier et al., 1994). The superoxide produced during reperfusion, however, appears to be rapidly converted to hydrogen peroxide (H_2O_2) . The H_2O_2 is in turn converted to hydroxyl radical through Fenton reaction that is stimulated by the presence of free iron accumulated during ischemia (Vanden Hoek et al., 1997). Production of both H₂O₂ and hydroxyl radical constitute the accelerated reactive oxygen species production in the first few minutes of reperfusion, also known as the reperfusion oxidant burst. Lipid peroxides and NADH/ NADPH reducing equivalents may accumulate during ischemia and could also help drive reactive oxygen species generation at reperfusion (Wang et al., 2006a). Other oxidants such as peroxynitrite are also produced, however, superoxide and H₂O₂/hydroxyl radical are generally regarded as the primary reactive oxygen species produced during ischemia/reperfusion (Li and Jackson, 2002). The highly reactive hydroxyl radical formed at reperfusion oxidizes cellular lipids, proteins and DNA, and is therefore very detrimental (Wang et al., 1999). Superoxide produced during ischemia also contributes to the overall ischemia/reperfusion injury. Thus, the reactive oxygen species generated during ischemia and reperfusion damage the cell. Antioxidant therapy aiming at mitigating ischemia/reperfusion injury, therefore, should effectively scavenge these specific reactive oxygen species produced during both ischemic and reperfusion phases so as to eliminate the oxidant injury.

In this study, we compared the oxidant scavenging capacity of 5 specific flavonoids (wogonin, baicalin, baicalein, catechin and procyanidin B₂, Fig. 1), the major natural antioxidant components frequently found in vegetables, fruits, tea, and herbs. The scavenging of superoxide and hydroxyl radical (the main reactive oxygen species associated with ischemia/reperfusion injury) with these flavonoids was measured. We then evaluated the protective efficacy in different treatment protocols: flavonoids were administered (1) as a 72-h pretreatment, (2) during ischemia and reperfusion, and (3) only at reperfusion, in an established chick cardiomyocyte model of ischemia/reperfusion. The correlation between the protective efficacy and the oxidant scavenging potency of the flavonoids was also discussed.

2. Materials and methods

2.1. Chemicals and solutions

The following chemicals were purchased: baicalin, baicalein, catechin, procyanidin B₂, propidium iodide, hydrogen peroxide, ammonium iron (II) sulfate hexahydrate, xanthine, diethylenetriaminepentaacetic acid (DTPA), 1,1-diphenyl-2picrylhydrazyl (DPPH), and spin-trap 5, 5-dimethyl *N*-oxide pyrroline (DMPO),

Fig. 1. Illustration of the chemical structures of flavonoids.

CH₃CN from Sigma Co. (St. Louis, MO); wogonin from Wako Pure Chemical Industries Ltd. (Japan); xanthine oxidase from Roche Applied Science (Indianapolis, IN). Balanced salt solution contained 117 mM NaCl, 4.0 mM KCl, 18 mM NaHCO₃, 0.8 mM MgSO₄, 1.0 mM NaHPO₄, 1.21 mM CaCl₂ and 5.6 mM glucose. To mimic the energy deprivation and hyperkalemic conditions in clinical ischemia, the ischemic balanced salt solution contained no glucose but 2-deoxyglucose 20 mM to inhibit glycolysis, and [K⁺] was raised to 8.0 mEq/L.

2.2. Electron spin resonance (ESR) spectrometric assay

The free radical scavenging by the five flavonoids (wogonin, baicalin, baicalein, catechin and procyanidin B₂) was analyzed using electron spin resonance (ESR) spectrometry (Varian E-109 X-Band, ESR Spectrometer. Varian Inc., Palo Alto, CA) in 3 radical producing systems: DPPH, superoxide (DMPO–xanthine/xanthine oxidase) and hydroxyl radical (DMPO–H₂O₂/FeSO₄) using a cell-free, *in vitro* system. The flavonoid concentration used for the different radicals was varied to demonstrate the maximal radical scavenging efficacy, but was kept constant per radical for comparison. Each experiment was repeated 3 times.

2.2.1. DPPH radical scavenging activity by ESR

The scavenging activity for DPPH, a stable radical, by flavonoids was estimated using ESR according to the method by Yamaguchi et al. (2000). Stock solution was made with flavonoids in CH₃CN and DPPH in ethanol. Reaction mixture contained 50 μ l test sample in phosphate buffered saline with 5% CH₃CN (volume/volume) and 50 μ l DPPH (0.5 mM) in ethanol. The final concentration of all flavonoids used was 50 μ M. ESR signal was recorded at room temperature, 2 min after sample mixing. Scavenging effect was determined by comparison with a solvent-treated control group.

2.2.2. Superoxide radical scavenging activity by ESR

The method of superoxide radical detection using spin trap DMPO were essentially the same as that described by Zhao et al. (2001). Superoxide radicals were produced by reaction of the xanthine/xanthine oxidase system and reacted with spin trap DMPO. Stock solutions were made with flavonoids in CH₃CN and the rest chemicals in phosphate buffered saline. The DMPO–OOH adducts were detected using ESR spectrometry. ESR signal was recorded 1 min after 20 μ l xanthine oxidase (0.5 U/ml, fresh made) was mixed with 20 μ l xanthine (5 mM), 20 μ l DTPA (0.5 mM), 20 μ l DMPO (250 mM), 15 μ l phosphate buffered saline and 5 μ l flavonoids in CH₃CN, with the final flavonoid concentration at 25 μ M.

2.2.3. Hydroxyl radical scavenging activity by ESR

The ESR spin-trapping technique was used to evaluate the hydroxyl radical scavenging activity (Noda et al., 1999). Hydroxyl radicals were generated by Fenton reaction and reacted rapidly with spin-trap DMPO. Stock solutions were made with flavonoids in CH₃CN and the rest in phosphate buffered saline. The resultant DMPO–OH adducts were detected using ESR spectrometry. ESR signal was recorded at room temperature,

10 min after 20 μ l FeSO₄ (1 mM, fresh made) was mixed with 20 μ l DMPO (150 mM), 20 μ l H₂O₂ (1 mM), 20 μ l EDTA (1 mM) and 20 μ l reference (CH₃CN) or samples (1 mM) in CH₃CN, with the final flavonoid concentration at 200 μ M.

2.3. Cardiomyocyte culture

Chick cardiomyocytes were isolated from 10-day chick embryos as previously described (Vanden Hoek et al., 1996). In brief, the hearts were removed and the ventricles were minced and enzymatically dispersed with 0.025% trypsin (Invitrogen, Grand Island, NY). The cell suspension containing trypsin inhibitor was centrifuged and then resuspended in the culture medium (54% balanced salt solution, 40% medium 199, 6% heat-inactivated fetal bovine serum, 100 U/ml penicillin, and 100 μ g/ml streptomycin). The cells were pipetted onto 25 mm glass coverslips with approximately 0.7×10^6 cells per coverslip, and incubated at 37 °C. The purity of the cardiomyocytes was confirmed by anti-myosin heavy chain monoclonal antibodies. All experiments were performed using the 4–5 day cultured cells, by which time a synchronously contracting layer of cells could be visualized with viability exceeding 95%.

2.4. Simulated ischemia/reperfusion

Cardiomyocyte culture medium was replaced with balanced salt solution and cardiomyocytes were incubated at 37 °C for 30 min. The cells were then placed into a hypoxic chamber (37 °C, 1% O₂, 20% CO₂ and 79% N₂) with balanced salt solution immediately changed to ischemia balanced salt solution preincubated overnight in the hypoxic chamber. The ischemic conditions were simulated and verified as before in our cardiomyocyte model (PO₂ 3–5 Torr, PCO₂ 144 Torr and pH 6.8) (Vanden Hoek et al., 1996). The coverslips were kept in the hypoxic chamber for 1 h, subsequently removed with an immediate change of the ischemic balanced salt solution to standard balanced salt solution, and incubated at 37 °C, 21% O₂ and 5% CO₂ for 3 h.

2.5. Therapeutic intervention

Five flavonoid antioxidants, wogonin, baicalin, baicalein, catechin and procyanidin B_2 (all at $25~\mu M$), were each tested in 3 different treatment protocols. For chronic treatment group, cells were treated with the tested flavonoids for 72 h and continued through the 1-h ischemia and 3-h reperfusion. For acute treatment groups, the tested flavonoids were added during both ischemia and reperfusion phases, or at reperfusion phase only. For ischemia/reperfusion control group, cells were subjected to 1-h simulated ischemia and 3-h reperfusion. In negative control group, cells were kept in balanced salt solution for 4.5 h. The protocols of all therapeutic interventions were shown in Fig. 2.

2.6. Viability assay

Cell viability was assessed with the exclusion fluorescent dye propidium iodide (5 μ M) measured at excitation 540 nm/

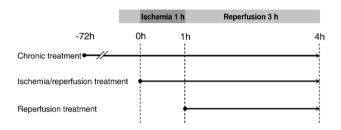


Fig. 2. Experimental protocols for cell viability assay in ischemia/reperfusion injury. Using an established chick cardiomyocyte model of simulated ischemia/reperfusion, tested flavonoids were: incubated since 72 h before ischemia and continued through ischemia/reperfusion (chronic treatment), treated during ischemia and reperfusion (ischemia/reperfusion treatment), or administered at reperfusion only (reperfusion treatment).

emission 590 nm using a Nikon TE 2000-U inverted phase/epifluorescent microscope (Bond et al., 1991). The dye was used throughout the entire experiment, exhibiting minimal toxicity in control cells even after a ten-hour exposure (Vanden Hoek et al., 1997). At the end of experiment, all cells on the coverslip were permeabilized with digitonin (300 μ M). Measurement of propidium iodide was done in 3 random fields of each coverslip at the end of 3-h reperfusion and after 1-h digitonin exposure. The fluorescence values were averaged. Percentage cell death (propidium iodide uptake) was expressed as the propidium iodide fluorescence relative to the maximal value seen after digitonin exposure (100%). Three to five coverslips were used per group with each coverslip representing n of one. Each experiment was repeated for at least 3 times in different batches of cells.

2.7. Statistical analysis

All results were presented as means±S.E.M. Comparison among different treatment groups was analyzed by one-way

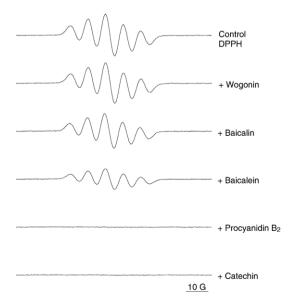


Fig. 3. Electron spin resonance (ESR) spectra of DPPH and DPPH radical-scavenging activity of different flavonoids: wogonin, baicalin, baicalein, procyanidin B_2 , and catechin at the same concentration of 50 μ M.

ANOVA with post-hoc examination by Tukey test. Difference was considered statistically significant if P<0.05.

3. Results

3.1. DPPH scavenging capacity of the flavonoids

Determining the quenching effect of the stable DPPH free radical is a rapid method to evaluate the radical scavenging activity of the tested compound. The scavenging capacity of the flavonoids was initially evaluated using the DPPH radical. As shown in Fig. 3. The DPPH scavenging capacity of the flavonoids at 50 μ M concentration was in the order of catechin (99±1%), procyanidin B₂ (92±5%), baicalein (55±4%), baicalin (17±4%), and wogonin (5±3%). Catechin and procyanidin B₂ exhibited excellent DPPH quenching effects with ESR signals almost completely eliminated.

3.2. Superoxide scavenging capacity of the flavonoids

Employing DMPO as the trapping agent, we evaluated the generation of superoxide using the xanthine/xanthine oxidase system and the superoxide scavenging effect of the flavonoids. As shown in Fig. 4, addition of the tested flavonoids to xanthine/xanthine oxidase plus DMPO system (final concentration 25 μ M) reduced the DMPO–OOH signals to different levels compared to that in control. The superoxide scavenging capacity of the five flavonoids was as follows: catechin (90±3%), baicalein (69±5%), procyanidin B₂ (68±5%), baicalin (42±4%), and wogonin (8±3%).

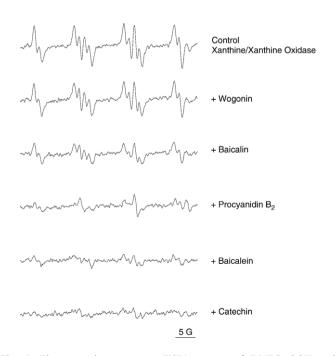


Fig. 4. Electron spin resonance (ESR) spectra of DMPO-OOH and superoxide radical-scavenging activity of different flavonoids: wogonin, baicalin, procyanidin B_2 , baicalein, and catechin at the same concentration of $25~\mu M$.

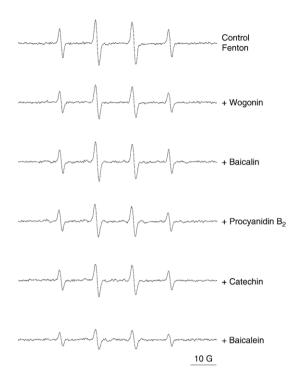


Fig. 5. Electron spin resonance (ESR) spectra of DMPO–OH and hydroxyl radical-scavenging activity of different flavonoids: wogonin, baicalin, procyanidin B_2 , catechin, and baicalein at the same concentration of 200 μ M.

3.3. Hydroxyl radical scavenging capacity of the flavonoids

The hydroxyl radical scavenging effect of the flavonoids was tested using a Fenton reaction system with DMPO as the trapping agent. Fig. 5 showed the typical 1:2:2:1 four-line ESR signal (with hyperfine splitting parameter $a^{\rm N}\!=\!a^{\rm H}\!=\!14.9$ G) of the DMPO–OH adducts. The addition of flavonoids (final concentration 200 μ M) resulted in differential quenching effects of the DMPO–OH adduct as demonstrated by reduction in the ESR signal intensity. In general, the scavenging effect of the flavonoids on hydroxyl radicals was weaker than those on DPPH or superoxide even with higher concentration (200 μ M): baicalein (56±4%), catechin (31±2%), procyanidin B2 (28±3%), wogonin (25±5%), and baicalin (15±3%). Only baicalein showed an ESR reduction rate of >50%.

The scavenging capacities of the 5 flavonoids in 3 different free radicals were summarized in Table 1.

Table 1
Oxidant scavenging potency of flavonoids

Oxidant scavenging (% reduction)	Wogonin	Baicalin	Baicalein	Catechin	$\begin{array}{c} Procyanidin \\ B_2 \end{array}$
DPPH	5±3	17±4	55±4	99±1	92±5
DMPO-OOH (superoxide)	8±3	42±4	69±5	90±3	68±5
DMPO-OH (hydroxyl radical)	25±5	15±3	56±4	31±2	28±3

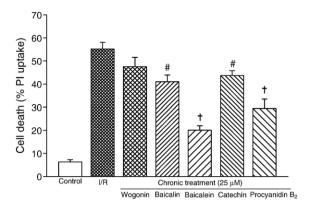


Fig. 6. Protective efficacy of flavonoids against cardiomyocyte ischemia/reperfusion injury in chronic treatment protocol. The tested flavonoids (25 μ M) were treated for 72 h prior to ischemia and continued through ischemia and reperfusion. ${}^{\#}P < 0.01$, ${}^{\dag}P < 0.001$ compared to ischemia/reperfusion (I/R).

3.4. Protective efficacy of the flavonoids against ischemia/reperfusion injury

We used an established chick cardiomyocyte model of simulated ischemia/reperfusion and tested the differential protective effects using 3 treatment protocols.

3.4.1. Chronic treatment

Cardiomyocytes subjected to 1-h ischemia followed by 3-h reperfusion resulted in $55.2\pm2.9\%$ cell death measured using the exclusion of propidium iodide (Fig. 6). In contrast, in cells incubated with balanced salt solution for 4.5 h, the percentage of cell death was $6.3\pm1.7\%$ (n=6, P<0.001), suggesting a significant detrimental effect of ischemia/reperfusion injury as we have previously described (Qin et al., 2004; Vanden Hoek et al., 1996). In chronic treatment group in which $25~\mu$ M flavonoids were administered for 72 h and through ischemia/reperfusion, most of the tested compounds showed protective effect against ischemia/reperfusion injury, except wogonin in which the cell death was not reduced significantly from the ischemia/reperfusion control ($47.5\pm4.1\%$ vs. $55.2\pm2.9\%$, n=6, P=NS; Fig. 6). The level of protection in the other four flavonoids varied, with baicalein exhibiting the best protective

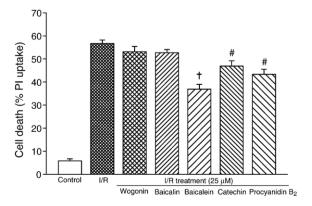


Fig. 7. Protective efficacy of flavonoids against cardiomyocyte ischemia/reperfusion injury in ischemia/reperfusion treatment protocol. The tested flavonoids (25 μ M) were administered during both ischemic and reperfusion phases. $^{\#}P$ <0.01, ^{+}P <0.001 compared to ischemia/reperfusion (I/R).

effect (cell death $20.1\pm1.9\%$, n=6, P<0.001 vs. ischemia/reperfusion control), followed by procyanidin B₂ (29.5 $\pm4.0\%$, n=6, P<0.001), baicalin (41.0 $\pm2.9\%$, n=6, P<0.01) and catechin (43.7 $\pm2.1\%$, n=6, P<0.01).

3.4.2. Treatment during ischemia and reperfusion

If the treatment with flavonoids was initiated from the beginning of ischemia and continued through reperfusion (ischemia/reperfusion treatment), the cardiomyocyte death was not attenuated as dramatically as with the chronic treatment. For example, the cell death for baicalein (25 µM) ischemia/reperfusion treatment was 36.9±2.1%, though lower than that of ischemia/reperfusion control (56.7 \pm 1.5%, n=6, P < 0.001; Fig. 7), was significantly higher compared to the chronic treatment (20.1 \pm 1.9%, n=6, P<0.001). Similar condition was also observed with procyanidin B2 treatment (ischemia/reperfusion treatment 43.3±2.6% vs. chronic treatment 29.5 \pm 4.0%, n=6, P<0.005). For catechin, the cell death in ischemia/reperfusion treatment ($46.9\pm2.3\%$) was lower than that of the ischemia/reperfusion control (P < 0.05; Fig. 7), but was not statistically significant from the chronic treatment group (43.7 \pm 2.1%, n=6, P=NS). For baicalin and wogonin, the cell death was not significantly reduced compared to that of the ischemia/reperfusion control (51.9±1.3% for baicalin and 53.1 ± 2.3% for wogonin vs. 56.7 ± 1.5% in ischemia/reperfusion, n=6, P=NS; Fig. 7).

3.4.3. Treatment during reperfusion only

For flavonoids given just at reperfusion (reperfusion treatment), the protective effect against ischemia/reperfusion injury was even less. While cell death after 3 h of reperfusion was still reduced in baicalein and procyanidin B_2 treatment groups (cell death $46.3\pm2.6\%$ and $47.8\pm3.7\%$, respectively, vs. $57.5\pm2.2\%$ in ischemia/reperfusion control, n=6, both P<0.05; Fig. 8), the level of protection was lower than those of chronic and ischemia/reperfusion treatments (Figs. 6 and 7). For catechin, baicalin and wogonin, no significant protective effect was seen with reperfusion treatment (cell death $54.6\pm1.7\%$, $54.6\pm2.3\%$ and $52.9\pm2.2\%$, respectively, n=6, P=NS

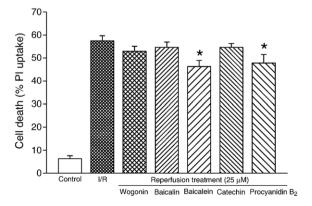


Fig. 8. Protective efficacy of flavonoids against cardiomyocyte ischemia/reperfusion injury in reperfusion treatment protocol. The tested flavonoids (25 μ M) were administered at reperfusion phase only. *P<0.05 compared to ischemia/reperfusion (I/R).

vs. ischemia/reperfusion control; Fig. 8). This suggests that flavonoid treatment at reperfusion only may not be effective in rescuing the ischemic tissues or cells from reperfusion injury.

4. Discussion

Flavonoids have attracted great attention in the prevention and treatment of cardiovascular diseases. With epidemiological evidence supporting that flavonoids reduce the risk of coronary artery disease (Hertog et al., 1993; Renaud and de Lorgeril, 1992), recent interest has been directed toward their therapeutic potential in the acute stages of the diseases (Rossi et al., 2003; Schultke et al., 2003). Our current study shows that five distinct flavonoids confer varied cardioprotection against ischemia/ reperfusion injury, depending on the treatment timing. The chronic treatment offers the best effect, while the reperfusion treatment has much lesser protective effect. We also found that the variant oxidant-scavenging capacity of flavonoids may partly contribute to their differential protective effects against ischemia/reperfusion injury.

4.1. Chemical characteristics determining the oxidant scavenging capacity

In general the oxidant scavenging capacity of flavonoids is related to the hydrogen donating ability and the stability of the phenoxyl radicals formed (Rice-Evans, 2004). Several factors determine the antioxidant capacity of flavonoids, such as the number and position of the phenolic hydrogens, the saturation status of the heterocyclic C ring, and any substitution of the phenolic hydrogens by glycoside or alkyl groups (Rice-Evans et al., 1996). For example, the ortho-dihydroxyphenolic structure of the B ring has been considered important for the oxidant scavenging activity (Fig. 1). The 3-OH group in the C ring, in conjunction with the adjacent 2,3-double bond and 4-oxo group, further potentiates the radical scavenging effect. The 3and 5-OH groups in the A ring also contribute in maximizing the radical scavenging potential. These structural characteristics affect the ability of electron delocalization from the aryloxyl radical on the B ring to the A ring and are thus responsible for the resonance effect of the aromatic nucleus that stabilizes the resulting phenoxyl radicals (Bors et al., 1990; Rice-Evans et al., 1996). Also, these characteristics contribute to the metal chelating activity, an important component of the flavonoids' antioxidant capacity (Sugihara et al., 1999).

In this study, catechin and procyanidin B₂ exhibited the best scavenging potency for DPPH and superoxide radicals as determined by ESR (Figs. 3, 4 and Table 1). This can largely be attributed to the abundant hydroxyl groups distributed in A, B and C rings that mediate hydrogen donation and electron delocalization. As procyanidin B₂ structurally is 2 epicatechin molecules linked from 4 position of the C ring in one molecule to 8 position of the A ring in the other (Fig. 1), its oxidant scavenging capacity is similar to that of catechin. Wogonin, on the other hand, was found to be poor in scavenging free radicals. This can probably be explained by the absence of hydroxyl groups in the B ring, even though there are 2,3 double bond and

4-oxo group in the C ring and two hydroxyl groups in the A ring. For baicalein, the absence of hydroxyl groups in the B ring is not different from that in wogonin (Fig. 1). However, the three hydroxyl groups in the A ring and the 2,3-double bond plus 4-oxo group in the C ring may account for its better oxidant scavenging capacity. For baicalin, as the 7-OH group in the A ring is substituted by glycoside (Fig. 1), the antioxidant potential is attenuated as compared to that of baicalein (Bors et al., 1990; Rice-Evans et al., 1996).

For hydroxyl radical, the scavenging capacity of the tested flavonoids was generally of lesser magnitude than those of DPPH and superoxide radicals, even if the concentration used for hydroxyl radical ESR experiment (200 µM) was significantly higher than those used for DPPH (50 µM) and superoxide (25 µM). This might in part explain why the protective effect is unsatisfactory if the flavonoids are administered only at reperfusion, since hydroxyl radical constitutes the major reactive oxygen species produced in this phase. One exception is baicalein, whose hydroxyl radical scavenging capacity is comparable to that for DPPH or superoxide (Table 1). Theoretically, the absence of OH groups in baicalein's B ring would attenuate the oxidant scavenging potency, making elimination of the hydroxyl radical poorer than other flavonoids with such a structure (e.g. catechin or procyanidin B₂). Nevertheless, the ESR result shows that, in contrast to DPPH or superoxide, baicalein's potency in hydroxyl radical scavenging is paradoxically better than the other flavonoids. This is probably due to the specific structural differences between baicalein and catechin that enables greater iron chelation and hydroxyl radical scavenging (Afanas'ev et al., 1989; Halliwell, 1995; Melidou et al., 2005). Iron chelation results in inhibition of hydroxyl radical production via Fenton reaction, thus contributing significantly to the overall antioxidant activity of the flavonoids.

4.2. Phase-dependent oxidant scavenging and protection against ischemia/reperfusion injury

We have previously reported in this chick cardiomyocyte ischemia/reperfusion model that superoxide generation increases during ischemia and declines rapidly in the first minutes of reperfusion (Becker, 2004; Becker et al., 1999). The oxidant burst at early reperfusion is composed mainly of H₂O₂ and hydroxyl radical (Li and Jackson, 2002), and is associated with accelerated cell death after reperfusion (Levraut et al., 2003; Vanden Hoek et al., 2003; Vanden Hoek et al., 1996). Thus there appears to be a phase-specific oxidant production during ischemia and reperfusion. In cardiomyocytes receiving chronic flavonoid treatment, the protective efficacy against ischemia/reperfusion injury is in the order of baicalein>procyanidin B₂>baicalin>catechin (Fig. 6). For ischemia/reperfusion treatment, the protective efficacy is baicalein>procyanidin B_2 >catechin (Fig. 7). Since these two protocols involve treatment with flavonoids during both ischemic and reperfusion phases, the results are consistent with their superoxide- and hydroxyl radical-scavenging capacity shown by ESR (Figs. 4, 5 and Table 1), thus protecting the cells from oxidant injury. The only exception is catechin, which exhibits high superoxide- and moderate hydroxyl radical-scavenging capacity but less satisfactory cytoprotection against ischemia/reperfusion injury at the tested concentration. The low lipid solubility of catechin, attributed to lack of 4-carbonyl group and 2,3-double bond in the C ring, may explain this discrepancy (Melidou et al., 2005). The ESR experiments are performed in the aqueous phase, while the ischemia/reperfusion experiments in cardiomyocytes are interactions with cell membrane and other lipophilic intracellular components. A differential antioxidant activity of flavonoids between aqueous and lipophilic phases has been demonstrated earlier. It is suggested that the lack of 4-carbonyl group and 2,3-double bond in the heterocyclic C ring of catechin may explain the weaker oxidant scavenging effect in the lipophilic phase, though such structure exhibits less important role in the antioxidant potency in aqueous system (Rice-Evans et al., 1996).

In cardiomyocytes treated with flavonoids during reperfusion only, the protective efficacy against ischemia/reperfusion injury is significant only with baicalein and procyanidin B₂. Since large quantity of hydroxyl radical is generated intracellularly in the early phase of reperfusion, the hydroxyl radicalscavenging capacity of the lipid-soluble flavonoids would become the determinant of the protective effect. Though the level of cytoprotection is less compared to that of ischemia/ reperfusion treatment and chronic treatment, the order of protective effect is consistent with the ESR results (Fig. 5 and Table 1). Again discrepancy is noted between catechin's protective effect and its hydroxyl radical-scavenging capacity, which is probably due to similar factors discussed above. Baicalein's high lipid-solubility, affording a rapid intracellular access (Saija et al., 1995), and its free iron-chelating ability (Afanas'ev et al., 1989; Halliwell, 1995), which in turn reduces hydroxyl radical formation, may explain its superior protective effect even when administered at reperfusion only (Shao et al., 2002). For procyanidin B₂, an increased capacity to prevent membrane injury due to availability of several hydroxyl group per molecule and an increased lipophilicity due to a compact conformation of the dimer, apart from its metal chelating activity, may explain the protective effect (da Silva Porto et al., 2003; Maffei Facino et al., 1996; Wang et al., 2000). Future studies measuring reactive oxygen species in situ using a more sensitive ESR technique need to be conducted to further support our observations.

4.3. Other possible protective mechanisms in chronic treatment strategy

Compared to acute treatment protocols, the protective effect against ischemia/reperfusion injury is most significant with chronic flavonoid treatment. This is consistent with previous studies and clinical observations suggesting that prevention is better than emergency treatment in face of acute oxidant injury (Hertog et al., 1997; Renaud and de Lorgeril, 1992). Administration of flavonoids prior to the ischemic insult would allow time for the flavonoids to accumulate intracellularly, which enables efficient oxidant scavenging from the beginning of ischemia.

Other factors such as activation of the endogenous antioxidant system may also contribute to the better protection seen in chronic treatment (Pataki et al., 2002). While 72-h pretreatment allows upregulation of the antioxidant proteins, which prepares the cell for an eventual ischemic insult, such activation is not possible in ischemia/reperfusion treatment or reperfusion treatment. Moreover, flavonoids have demonstrated potentials in modulating a number of second messengers and other intracellular processes that are involved in oxidant stress and cell death (Rice-Evans, 2001, 2004; Williams et al., 2004). Preadministration of flavonoids accomplishes these before the ischemia/reperfusion insult, thus preparing the cell in advance. Further work is needed for better understanding the alterations in endogenous antioxidant enzymes and chemicals that may be responsible for protection from oxidant injury in the chronic treatment model.

In conclusion, flavonoids' antioxidant protection against ischemia/reperfusion injury is dependent on the oxidant scavenging capacity for the reactive oxygen species generated during ischemia/reperfusion, the timing of flavonoid treatment, and the membrane permeability to flavonoids. The flavonoidspecific oxidant-scavenging capacity may simply be explained on the basis of their chemical structures. However, the cytoprotective effects of the flavonoids are probably determined, apart from their scavenging capacity, by intracellular bioavailability of the flavonoids and their potential pharmacodynamic effects within the cardiomyocytes. The flavonoids with high scavenging capacity for oxidant species produced during ischemia (superoxide) and reperfusion (hydroxyl radical), in addition to possessing higher lipid solubility, exhibit best protective efficacy. Thus, when administered as an acute treatment, only those flavonoids with favorable chemical characteristics ensuring rapid intracellular access and scavenging of relevant toxic reactive oxygen species will be efficacious in protecting against ischemia/reperfusion injury.

Acknowledgements

This work was supported in part by NIH grants HL68951, HL72734, AT002176, AT003255 and AT01575-01.

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